

Mechanisms of Heat Transfer

- We want to know the rate at which energy is transferred
- There are various mechanisms responsible for the transfer:
 - **Conduction**
 - **Convection**
 - **Radiation**

Conduction

- The transfer can be viewed on an atomic scale
 - It is an exchange of energy between microscopic particles by collisions
 - The microscopic particles can be atoms, molecules or free electrons
 - Less energetic particles gain energy during collisions with more energetic particles
- Rate of conduction depends upon the characteristics of the substance

Conduction example

- The molecules vibrate about their equilibrium positions
- Particles near the heat source vibrate with larger amplitudes
- These collide with adjacent molecules and transfer some energy
- Eventually, the energy travels entirely through the pan



© 2004 Thomson/Brooks Cole

Conduction

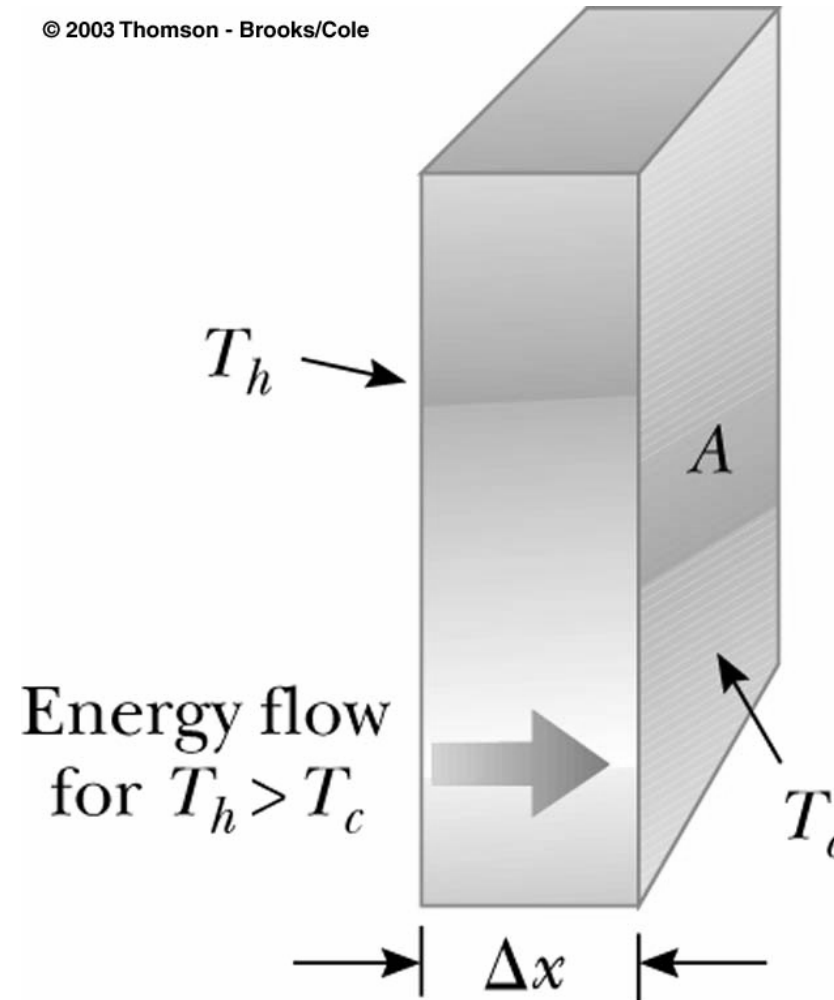
- **Metals are good conductors**
 - They contain **large numbers of electrons** that are relatively free to move through the metal
 - They can transport energy from one region to another
- Poor conductors include asbestos, paper, and gases
- **Conduction can occur only if there is a difference in temperature between two parts of the conducting medium**

Conduction equation

- The slab at right allows energy to transfer from the region of higher temperature to the region of lower temperature
- The rate of transfer is given by:

$$\mathcal{Q} = \frac{Q}{\Delta t} = kA \left| \frac{dT}{dx} \right|$$

© 2003 Thomson - Brooks/Cole



Conduction equation: explanation

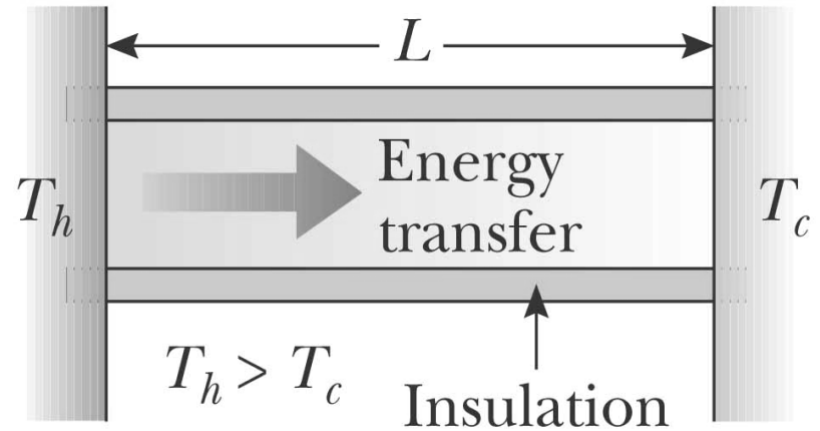
- A is the cross-sectional area
- x is the thickness of the slab
 - Or the length of a rod
- P is in Watts when Q is in Joules and t is in seconds
- k is the *thermal conductivity* of the material
 - Good conductors have high k values and good insulators have low k values

Temperature Gradient

The quantity $|dT / dx|$ is called the **temperature gradient** of the material

- It measures the rate at which temperature varies with position

For a rod, the temperature gradient can be expressed as:



© 2004 Thomson/Brooks Cole

$$\left| \frac{dT}{dx} \right| = \frac{T_h - T_c}{L}$$

Rate of Energy Transfer in a Rod

- Using the temperature gradient for the rod, the rate of energy transfer becomes:

$$\dot{Q} = kA \left(\frac{T_h - T_c}{L} \right)$$