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PH4211 Statistical Mechanics 2011
Outline Answers.

Q1 a) λ is the de Broglie wavelength corresponding to a free particle of momentum p , corresponding to a kinetic energy E corresponding to the thermal energy kT (to within a factor of π).

③

b) Since Z is the sum of Boltzmann factors - i.e. it is related to possibilities, then the possibilities for independent events multiply, so the Z s would multiply. 2

Since the free energy comes from the logarithm of Z , then to be consistent with extensivity of energy, the Z s must multiply. 2

So generally, for independent systems: $Z = Z_1 Z_2 Z_3 \dots$

And so if all N systems have the same mathematical expression for Z , then $Z = Z^N$. 1

③

c)
$$F = -kT \ln Z = -NkT \ln z$$
$$= -NkT \ln \left[V \left(\frac{m k T}{2 \pi \hbar^2} \right)^{3/2} \right]$$

③

d) Although the N is needed for extensivity, since it is said that the total free energy F is proportional to N , this is not enough. The problem is that the argument of the \ln is

proportional to V . True, the argument is dimensional, but this is not enough.

Extensivity requires

$$\frac{F}{N} \rightarrow \text{limit } f \text{ as } N, V \rightarrow \infty$$

while $\frac{N}{V}$ remains constant.

In the existing Brn you don't get a const value, instead $F \rightarrow \infty$!!

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e) Mathematically the problem would be solved by dividing the V into h^3 by a dimensional quantity proportional to N .

If we had some αN then the α would contribute an additive constant to F . So not a problem classically. 2

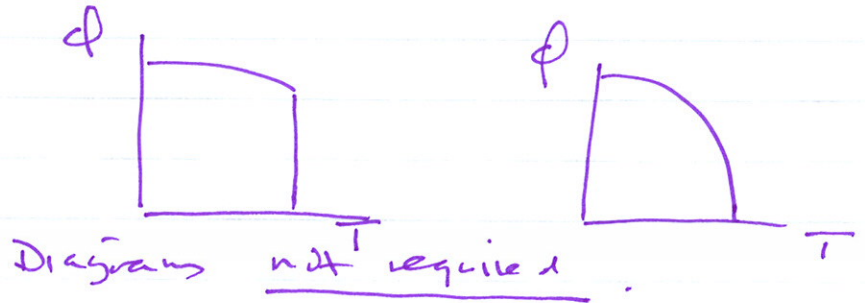
From the quantum perspective the particle indistinguishability leads to a factor $\frac{1}{N!} \sim - (N \ln N - N)$ by Stirling's approximation. This gives the required correction to the argument of the \ln . Write $N = N \ln e$ 2 so that $\frac{1}{N!} \sim - N \ln(N/e)$.

gives

$$F = -NKT \ln \left[\frac{V}{N} \left(\frac{mKT}{2\pi h^2} \right)^{3/2} e \right]$$

5

Q2 a) Order parameter is a quantity giving a measure of the order in the ordered phase. It may be a scalar, vector, complex, tensor. In a 1st order transition the order parameter goes to zero discontinuously at the transition. In a 2nd order transition the order parameter goes to zero continuously.



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b) Order parameter for ferroelectric transition is the electric polarization \vec{P} , although it will only point along certain axes related to the crystal lattice structure.

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c) Series is terminated at the lowest (even) power that is capable of exhibiting the correct generic behavior at the transition. It must be an even power so that the free energy is banded, i.e. system is stable.

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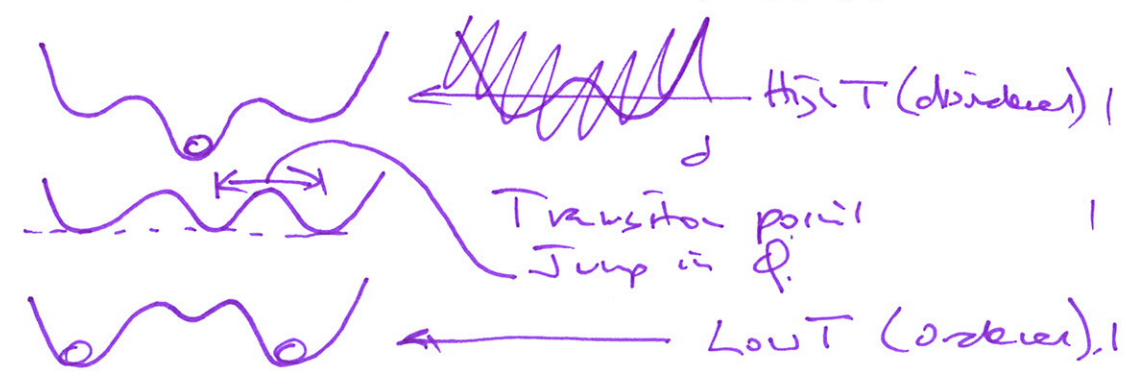
There are no odd powers in the expansion because of the symmetry - System has inversion symmetry - if we replace \vec{P} by $-\vec{P}$ then it must ^{free energy}

be unchanged. 2
 The free energy is a scalar. If
 we take \vec{P} is the odd parameter
 and if there are no other directions
 in the problem then the only way
 to make a scalar out of \vec{P} is
 to dot it with itself. So we
 only have powers $A P^2$ 2

4



This is the general form for a 6th
 order even polynomial.
 When the central minimum is below
 the other two we are in the
 disordered (High-T) state.
 But when the side minima fall
 below the central one there will
 be a discontinuous jump of ϕ
 - i.e. a 1st order transition



4

e) For the transition to be 1^{\rightarrow} order we require $F_4 < 0$ or $F_6 > 0$
 If F_4 becomes positive then we no longer need the F_6 term.
 With $F_4 > 0$ the transition is 2^{\rightarrow} order. (3)

f) At the transition the three minima are at the same height. Let us put $F_0 = 0$ so the central minimum is at $F = 0$ as so the outer ones are zero as well.
 Then $F_2 \phi^2 + F_4 \phi^4 + F_6 \phi^6 = 0$
 The fact that these are also minima means that $\frac{dF}{d\phi} = 0$. i.e. $2F_2 \phi + 4F_4 \phi^3 + 6F_6 \phi^5 = 0$
 So $\phi = 0$ or $F_2 + 2F_4 \phi^2 + 3F_6 \phi^4 = 0$

The Simultaneous eqns:

$$\begin{aligned} F_2 + F_4 \phi^2 + F_6 \phi^4 &= 0 \\ F_2 + 2F_4 \phi^2 + 3F_6 \phi^4 &= 0 \end{aligned}$$

Eliminate F_2
 $\rightarrow F_4 + 2F_6 \phi^2 = 0$ or $\phi = \sqrt{-\frac{F_4}{2F_6}}$

But F_6 is +ve. F_4 is -ve.
 so this gives the $\phi \neq 0$ roots.

So the jump in ϕ is from 0 to $\sqrt{-\frac{F_4}{2F_6}}$
 or $\Delta\phi = \sqrt{-\frac{F_4}{2F_6}}$ 3

F_4 goes to zero as the transition becomes 2^{\rightarrow} order. So from the expansion for $\Delta\phi$ we see that $\Delta\phi \rightarrow 0$ is expected. (4)

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a) Virial expansion is an expansion in density $\frac{N}{V}$ — so like order terms may be increasingly neglected as the density gets smaller.

②

b) Re-arrange $VdU = \frac{P}{KT} V^2 - \frac{a}{KT} \frac{N^2}{V}$

$$\begin{aligned} \text{or } \frac{P}{KT} &= \frac{N}{V} \left(\frac{1}{1 - \frac{Nb}{V}} \right) - \frac{a}{KT} \frac{N^2}{V^2} \\ &= \frac{N}{V} \left(1 + \frac{N}{V} b + \dots \right) - \frac{a}{KT} \frac{N^2}{V^2} \\ &= \frac{N}{V} + \left(\frac{N}{V} \right)^2 \left(b - \frac{a}{KT} \right) + \dots \end{aligned}$$

So $B_2 = b - \frac{a}{KT}$

④

c) From the plot the $1/T = 0$ intercept is approx $17 \text{ cm}^3/\text{mol}$. So $b = 17 \text{ cm}^3/\text{NA}$
 or $2.8 \times 10^{-23} \text{ cm}^3 = \frac{2}{3} \pi \sigma^3$ so ~~XXXXXXXXXXXX~~
 or $\sigma^3 = 1.35 \times 10^{-23} \text{ cm}^3$ so $\sigma = 2.38 \times 10^{-8} \text{ cm}$
 $= 2.38 \times 10^{-10} \text{ m}$

Step A plot is $\sim 100 b$

c) From the plot of the B_2 data to $1/T=0$ intercept gives $b = 17 \text{ cm}^3/\text{mol}$
 i.e. $b = 17 \text{ cm}^3/\text{NA} = \underline{2.8 \times 10^{-23} \text{ cm}^3}$

or $\underline{b = 2.8 \times 10^{-29} \text{ m}^3}$ 2

Step A curve = $-\alpha/k = -100b$

So $\alpha = 100b \times k = 2.8 \times 10^{-23} \times 1.4 \times 10^{23}$
 $\underline{\alpha = 3.9 \times 10^{-50} \text{ J m}^3}$ 2

Important approximation/assumption is to draw the line through the high-temp data where the quantum effects are negligible. 2

d) ϵ is the depth of the attractive well (energy) and a is the hard-core diameter. 2

e) b comes from the hard-core value so it is related to $\frac{4}{3}\pi a^3$
 a comes from $a \frac{N^2}{V^2} = N \frac{d\langle E \rangle}{dV}$
 $\text{and } \langle E \rangle \sim \int_0^\infty 4\pi r^2 U(r) dr$
 which then gives us $a = \frac{4}{3}\pi a^3 \epsilon$
 for the L-J 6-12 potential for $U(r)$. 3

$\sigma = \left(\frac{3b}{4\pi}\right)^{1/3} \rightarrow 2.38 \times 10^{-8} \text{ cm} \text{ or } 2.38 \times 10^{-10} \text{ m}$

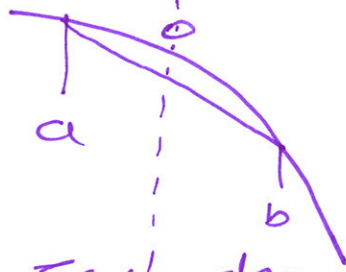
$\epsilon = \frac{a}{2b} = 7 \times 10^{-22} \text{ J} \left(\frac{\epsilon}{k} \approx 50 \text{ K}\right)$ 3

- 4 a) Minimize Helmholtz @ const T, V !
Minimize Gibbs @ const T, P !

2

- b) Order parameter must be constrained !

If free energy is concave.



Then can drop off curve to the straight line. So here reduced the free energy and here some material in state a and some in state b rather than all in state o.

Populations determined by minimizing the mean value of the order parameter what it is spontaneous (constrained) to be.

2

- c) Equilibrium state is determined by minimizing the free energy so must continue dropping below the curve until you can go no further. — Ultimately the free energy must look like this:

3



Here the line gives the best free energy — equilibrium state.

3

d) Critical point is the "Zwickpunkt"
 A the two phases. — when
 the a and the b phase go
 together.

This is when the "wiggle" in
 the free energy disappears.
 It corresponds to the highest
 temperature A the transition.

2

- e) N: n^o A particles / site
- S: n^o A neighbors
- X: Concentration A one species
- k: Boltzmann's const
- T: temperature.

First term is energy term
 second term is entropy (A 2-state system)

2

f) Coexistence curve corresponds to minimizing
 free energy:

$$\frac{dF}{dx} = NSE(1-2x) - NkT \ln\left(\frac{1-x}{x}\right) = 0$$

$$\Rightarrow T_{CS} = \frac{SE(1-2x)}{k \ln\left(\frac{1-x}{x}\right)}$$

Symmetry $\Rightarrow x_c = 1/2$

take limit at T_{CS}
 $\Rightarrow T_c = \frac{1SE}{2k}$

3

g) $x_c = \frac{1}{2}$ by symmetry

2

h) Similar slope — but alloy curve is symmetric
 while fluid curve is skewed.
 β critical exponent for both is $1/3$

3

3. (a) The virial expansion for a gas is given by

$$\frac{p}{kT} = \frac{N}{V} + \left(\frac{N}{V}\right)^2 B_2 + \left(\frac{N}{V}\right)^3 B_3 + \dots;$$

here the B_n are the virial coefficients and the other symbols have their usual meaning. Under what conditions may the higher-order terms be ignored?

[2]

- (b) The van der Waals equation of state is written, in conventional form, as

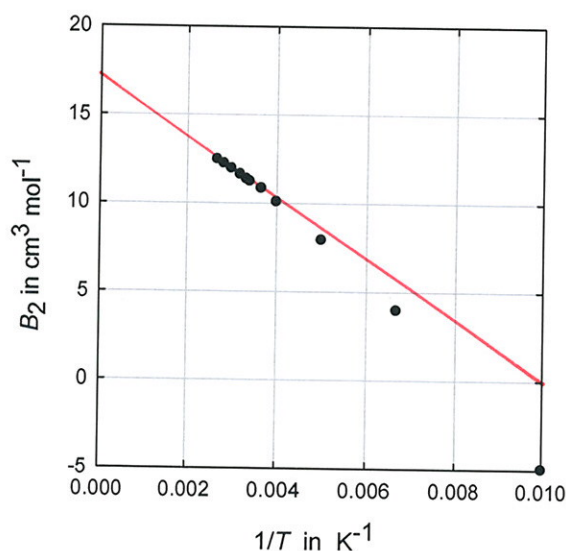
$$\left(p + a \frac{N^2}{V^2}\right)(v - Nb) = NkT.$$

Show that the second virial coefficient corresponding to this equation of state is

$$B_2 = b - \frac{a}{kT}.$$

[4]

- (c) Measurements of the second virial coefficient of neon are shown in the figure below.



Estimate the values of the van der Waals parameters a and b . Be sure to state your units. Explain any approximations or assumptions.

[6]

- (d) The interaction between neon atoms is conventionally represented by the Lennard-Jones 6-12 potential

$$U(r) = \varepsilon \left\{ \left(\frac{\sigma}{r}\right)^{12} - \left(\frac{\sigma}{r}\right)^6 \right\};$$

what are ε and σ in this expression?

[2]

- (e) Outline briefly the arguments by which one makes the identification

$$a = \frac{4}{3} \pi \sigma^3 \varepsilon \quad \text{and} \quad b = \frac{2}{3} \pi \sigma^3$$

and hence estimate ε and σ for argon.

[6]