# **UNIVERSITY OF LONDON**

## **MSci EXAMINATION 2010**

For Students of the University of London

# **DO NOT TURN OVER UNTIL TOLD TO BEGIN**

## PH4211: STATISTICAL MECHANICS

Time Allowed: **TWO AND A HALF** hours

Answer **THREE** questions only

Please answer each question on a separate page.

Approximate part-marks for questions are given in the right-hand margin.

No credit will be given for attempting any further questions.

College Calculators are provided.

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## GENERAL PHYSICAL CONSTANTS

Permeability of vacuum	$\mu_0$	=	$4\pi \times 10^{-7}$	$H m^{-1}$
Permittivity of vacuum	$\epsilon_0$	=	$8.85 \times 10^{-12}$	F m <sup>-1</sup>
	$1/4\pi\epsilon_0$	=	$9.0 \times 10^{9}$	${ m m}{ m F}^{-1}$
Speed of light in vacuum	С	=	$3.00 \times 10^{8}$	-1 m s
Elementary charge	е	=	$1.60 \times 10^{-19}$	С
Electron (rest) mass	m <sub>e</sub>	=	$9.11 \times 10^{-31}$	kg
Unified atomic mass constant	$m_{_{ m U}}$	=	$1.66 \times 10^{-27}$	kg
Proton rest mass	$m_{ m p}$	=	$1.67 \times 10^{-27}$	kg
Neutron rest mass	m <sub>n</sub>	=	$1.67 \times 10^{-27}$	kg
Ratio of electronic charge to mass	<i>e/m</i> <sub>e</sub>	=	$1.76\times10^{11}$	$C kg^{-1}$
Planck constant	h	=	$6.63 \times 10^{-34}$	Js
	$\hbar = h/2\pi$	=	$1.05 \times 10^{-34}$	Js
Boltzmann constant	k	=	$1.38 \times 10^{-23}$	J K <sup>-1</sup>
Stefan-Boltzmann constant	σ	=	$5.67 \times 10^{-8}$	$W m^{-2} K^{-4}$
Gas constant	R	=	8.31	$\mathbf{J} \operatorname{mol}^{-1} \mathbf{K}^{-1}$
Avogadro constant	$N_{_{\rm A}}$	=	$6.02 \times 10^{23}$	mol
Gravitational constant	G	=	$6.67 \times 10^{-11}$	$Nm^{2}kg^{-2}$
Acceleration due to gravity	g	=	9.81	-2 m s
Volume of one mole of an ideal gas at STP		=	$2.24 \times 10^{-2}$	m <sup>3</sup>
One standard atmosphere	$P_{0}$	=	$1.01 \times 10^5$	N m <sup>-2</sup>

## MATHEMATICAL CONSTANTS

$$e \cong 2.718$$
  $\pi \cong 3.142$   $\log_{e} 10 \cong 2.303$ 

#### PART MARKS

1. The relationship between statistical mechanics and thermodynamics may be understood in different ways. According to Landau 'All the concepts and quantities of thermodynamics follow most naturally, simply and rigorously from the concepts of statistical physics', while Einstein's view was that thermodynamics was a fundamental discipline whose results were independent of the particular microscopic models used in statistical mechanics.

Write an essay discussing these contrary views; give examples supporting the Landau view, and examples supporting Einstein's view. Discuss the extent to which the work of both Landau and Einstein were actually more in line with the philosophy of the other. You should include mention of *emergence* as the key concept that resolves the Landau-Einstein conflict.

			PART MARKS
2.		Landau theory of phase transitions one expands an appropriate free energy ers of the order parameter, to a finite number of terms.	
	(i)	What is meant by the term <i>order parameter</i> ?	[3]
	(ii)	Why is the Landau expansion terminated, and what determines the highest power in the expansion?	[4]
	(iii)	In the case of the ferromagnetic transition and the ferroelectric transition there are no odd-power terms in the expansion. Why is this?	[2]
	(iv)	In the ferroelectric case the Landau expansion has terms up to sixth power. Show, with the aid of diagrams, why such an expansion can describe a first-order transition.	[3]
	(v)	This expansion will also allow a second-order transition. Discuss what requirements this puts on the sign of the terms in the Landau expansion and explain the occurrence of the second-order transition with the aid of diagrams.	[5]
	(vi)	Explain, in general terms, how the heat capacity in the vicinity of a phase transition can be understood from the perspective of the Landau theory. In particular, make the connection between the order of the transition and the existence or absence of latent heat.	[3]

## TURN OVER

#### PART MARKS

[3]

3. (a) The interaction between the molecules of a gas may be modelled by the [4] hard core potential:

$$U(r) = \infty \qquad r < \sigma$$
$$= 0 \qquad r > \sigma$$

Make a labelled plot of this function and discuss the extent to which this potential gives a realistic description of such interactions.

(b) The second virial coefficient of a gas is related to the inter-atomic [4] potential by

$$B_2(T) = -2\pi \int_0^\infty r^2 \left[ e^{-U(r)/kT} - 1 \right] \mathrm{d}r \,.$$

Show that for the *hard core* interaction the second virial coefficient is given by

$$B_2(T)=\frac{2}{3}\pi\sigma^3.$$

Demonstrate how this may be interpreted as an 'excluded volume' [4]

(c) The Sutherland potential is specified by:

$$U(r) = \infty \qquad r < \sigma$$
$$= -\varepsilon \left(\frac{\sigma}{r}\right)^{6} \qquad r > \sigma$$

Sketch and label this potential. How realistic is this model as an approximation of a real inter-atomic interaction i) at short distances, and ii) at large distances?

The figure shows the second virial coefficient corresponding to the hard sphere potential, the Sutherland potential and the Lennard-Jones 6-12 potential. Discuss and explain the differences and similarities.

## **TURN OVER**

	PART MARKS
The free energy of mixing of a binary alloy may be written as	[3]
$F_{\rm m} = Nsx(1-x)\varepsilon + NkT\left\{x\ln x + (1-x)\ln(1-x)\right\}.$	
Explain the meaning of the various terms and the structure of this equation.	
What are the equilibrium (stable) states of this system i) at high temperatures, and ii) at low temperatures?	[5]
How does one determine the equilibrium state of this system?	
Explain why one may use the free energy of mixing rather than the full free energy.	
Make a clearly labelled sketch of the transition between the homogeneous and the inhomogeneous phases in the $x - T$ plane. Indicate on this curve how the system evolves from a homogeneous high-temperature phase to an inhomogeneous low-temperature phase.	[2]
Show that this system has a critical temperature given by	[4]
$T_{\rm c} = \frac{s\varepsilon}{2k}$	
and state the value of the critical concentration $x_c$ .	
Explain why the point ( $x_c$ , $T_c$ ) is called a critical point.	[2]
Sketch the heat capacity for such a system as a function of temperature in the vicinity of the phase transition (i) for a mixture of critical concentration $x_c$ , and (ii) for a mixture of concentration $x \ll x_c$ .	[4]
These heat capacity signatures indicate there is <i>no</i> latent heat. What feature of the heat capacity indicates this?	
It appears paradoxical that this first-order transition involves no latent heat. Why is this?	

5.	(a)	In an isolated system the probability $P(x)$ that a fluctuating quantity has the value x is given by the Einstein expression	PART MARKS [3]
		$P(x) \propto e^{S(x)/k}$	
		where $S$ is the entropy and $k$ is Boltzmann's constant. Give a justification for this formula.	
		Sketch and explain the functional form of $S(x)$ in the vicinity of the mean value $\langle x \rangle$ .	[3]
		Hence discuss qualitatively why the fluctuations in $x$ may follow a normal distribution.	[3]
	(b)	The autocorrelation function of the fluctuating quantity $x$ is given by	[4]
		$G(t) = \langle x(t)x(0) \rangle$	
		Where the angled bracket indicates an average. Explain how $G(t)$ describes the 'average' decay of the fluctuations. (Here the mean value $\langle x \rangle$ is taken to be zero).	
		The correlation time of the fluctuations is defined by	[4]
		$\tau_{\rm c} = \frac{1}{G(0)} \int_0^\infty G(t) dt  .$	

What feature of the fluctuating quantity x(t) does the correlation time describe? Give an explanation of this.

(c) Now consider the case where the fluctuations in x are normally distributed [3] (and again the mean value  $\langle x \rangle$  is taken to be zero). Then P(x) is given by

$$P(x) = \frac{1}{\sqrt{2\pi \langle x^2 \rangle}} e^{-x^2/2 \langle x^2 \rangle}.$$

Show how the probability distribution function P(x) is related to the autocorrelation function G(t).

### **END**